

Problem 15 Water from the Air

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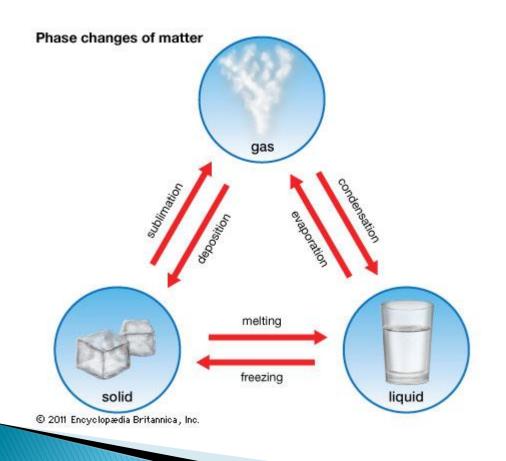


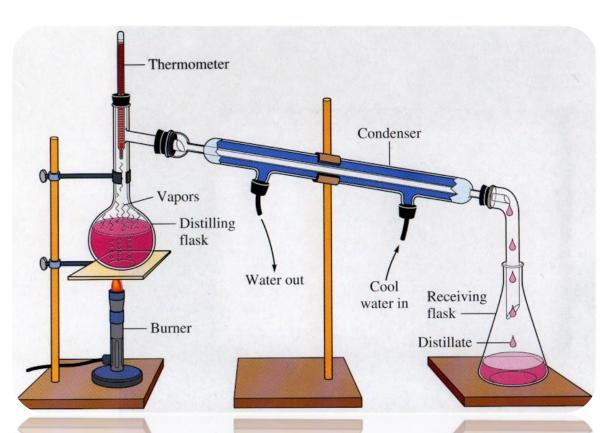
WATER FROM THE AIR

Design and construct a device allowing collection of water by condensing moisture from air. Determine if the water obtained with your device is suitable for drinking. What amount of water is possible to collect during one Science Fight?

WHAT IS CONDENSATION?

Condensation is the process in which water vapours turn into liquid water





THEORETICAL MAXIMUM

The theoretical maximum can be reached only if the dry ice gains all the heat which water vapours have radiated

$$M_{CO2} * r_{CO2} = M_{H20} * r_{H2O}$$

Where:

 M_{CO2} is the mass of the dry ice r_{CO2} is the specific sublimation heat of dry ice M_{H20} is the mass of the water collected r_{H2O} is the latent heat of water

EXPERIMENTAL SETUP

The device is made from a plastic cylinder and a computer fan on top





OPERATION

To operate the device you need to fill the cylinder with dry ice







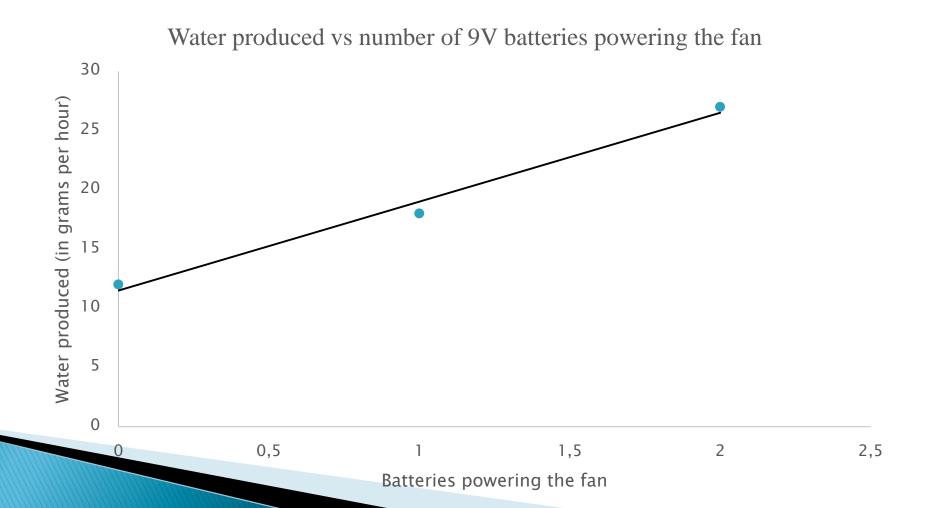
HOW AND WHY IT WORKS

When the cylinder gets very cold, water vapour in the air turns into water and then, almost immediately, into water crystals

$$\begin{array}{c|c} -Q_1 \\ \hline \end{array}$$
 Liquid
$$\begin{array}{c|c} -Q_2 \\ \hline \end{array}$$
 Solid

EFFICIENCY

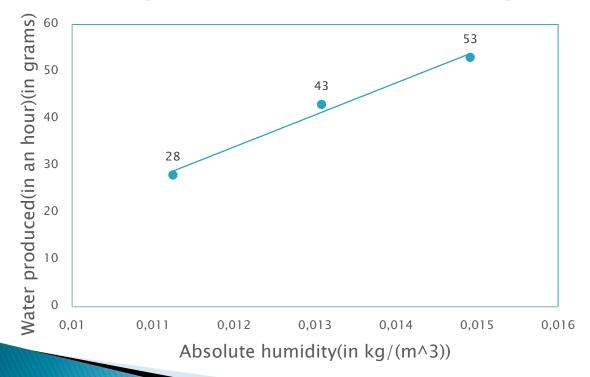
The productivity depends highly on the water vapour concentration in the air



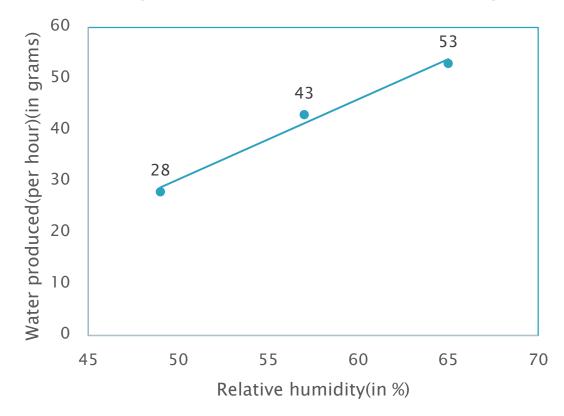
EXPERIMENTAL RESULTS

The results follow a straight line very accurately, which was expected

Water produced vs absolute humidity



Water produced vs Relative humidity



DRINKABILITY

The water collected with the device is suitable for drinking because it is basically distilled water and the device has been previously cleaned



CONCLUSION

Trying to extract water from the air by condensation is a process whose results are measurable and repeatable

▶ The water obtained by condensation is suitable for drinking

THANK YOU FOR THE ATTENTION